Crysta'days November 8 2023

Introduction to Secoya Crystallization Technology



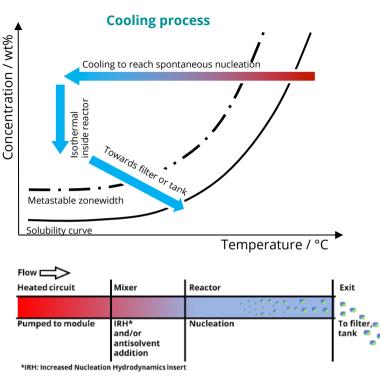
Secoya Crystallization Technology Where we are now



Several crystallization modes Cooling crystallization Antisolvent co-flow mode Antisolvent frontal mode

High control on size and dispersity

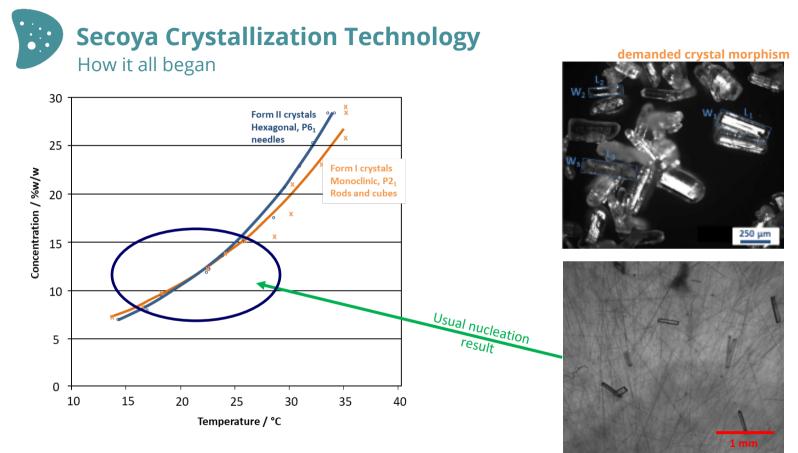
From nm to 400 µm size Low polydispersity (span < 2) Excellent flowability (non charged particles)



Rimez et al., Crystal Growth & Design (2018)

Any parameter influencing the nucleation can be optimized and control to allow a final crystal size selection adapted to the application, in one step and with a low size distribution.





NH2



Rimez et al. Cryst Growth Des 2018

3

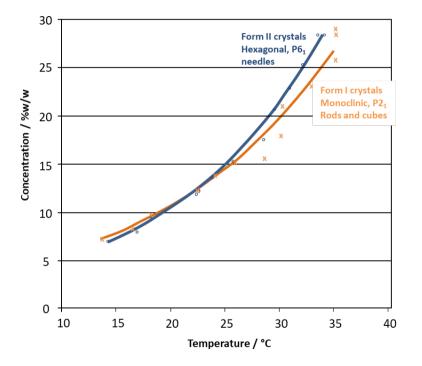
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NH2



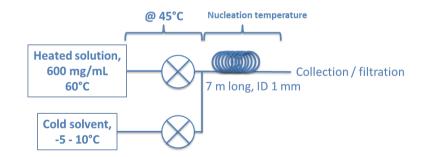
Secoya Crystallization Technology

How it all began



Evolution (lots of it can be found in PhD J. Conté):

- Impinging jet crystallization using antisolvent at various temperatures (toluene, hexane, etc.)
- Any interaction with foreign bodies
- Droplet crystallization in microfluidic setting
- Impinging jet mixing with solvent IPAC at room temperature
- Impinging jet mixing with cooled solvent IPAC





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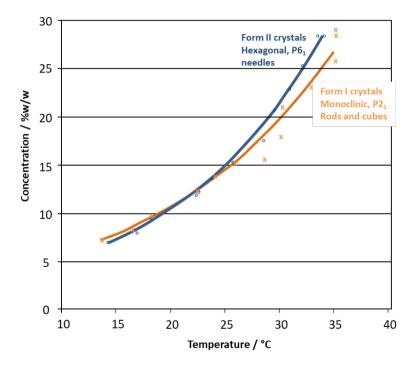
NH2



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Secoya Crystallization Technology

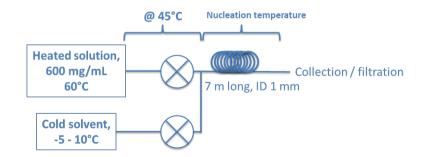
How it all began



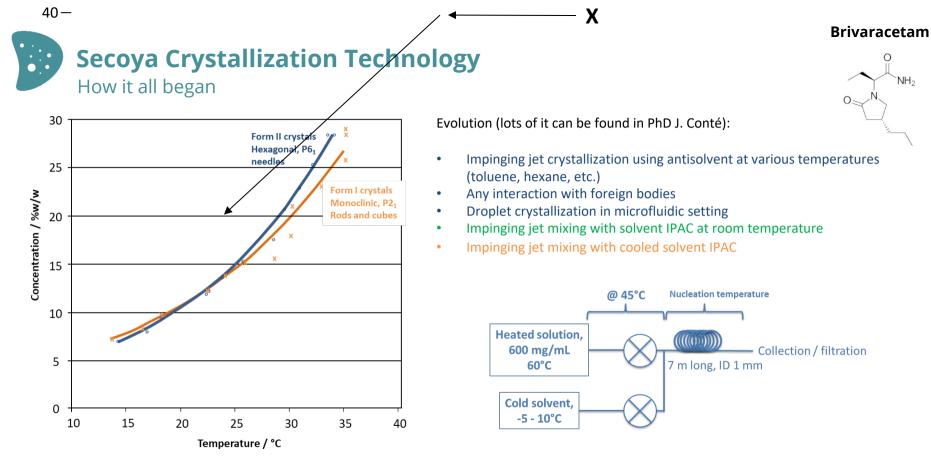
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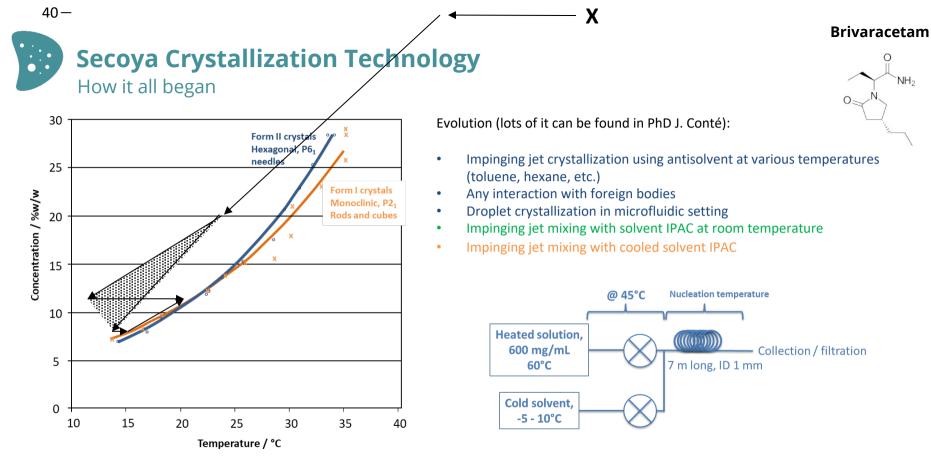
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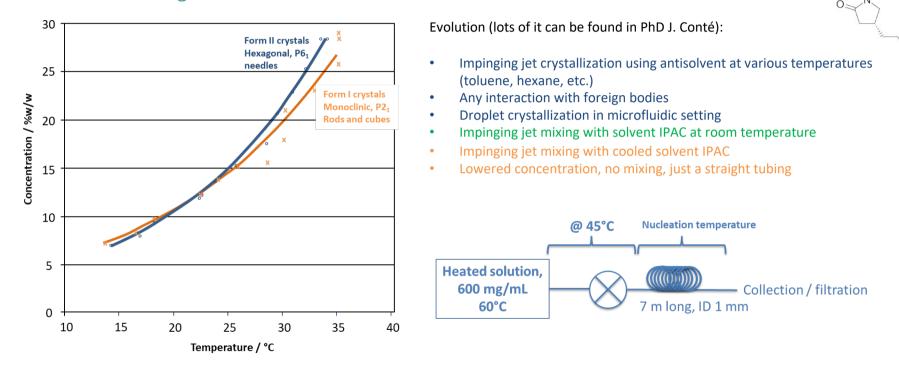
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Secoya Crystallization Technology

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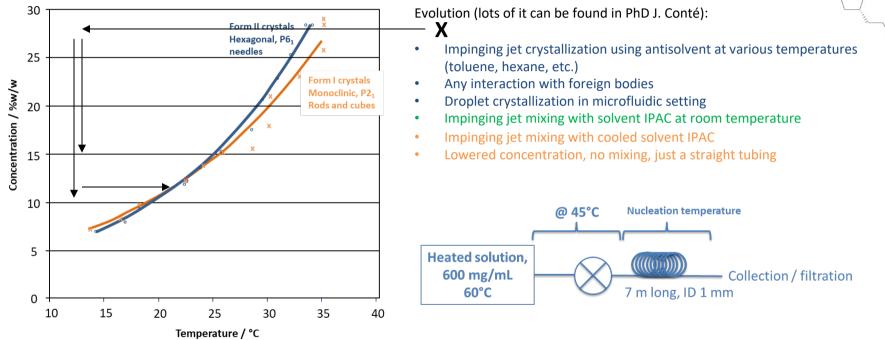
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NH2



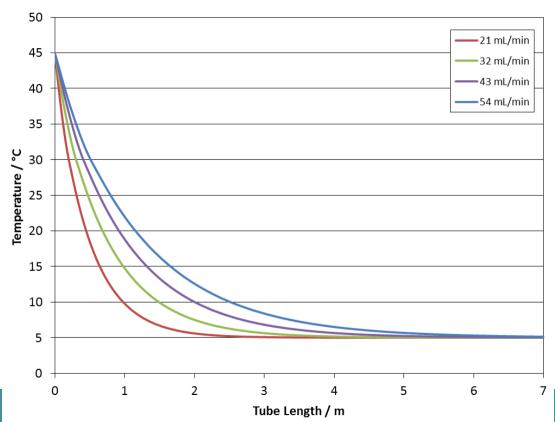






Secoya Crystallization Technology

Residence time limitation?



Residence time in tubing

Flow mL/min	Velocity cm/s	Residence time s
21	45	16
32	68	10
43	91	8
54	115	6



Secoya Crystallization Technology

Residence time limitation?



Concentration	Flow	Velocity	Residence time	Temperature bath	Pressure	Crystallization result	Crystal sizes
mg/mL	mL/min	cm/sec	sec	°C	bar		μm
600	21	45	16	0	8	Cubic	175± 100
600	32	68	10	0	12	Cubic	121 ± 33
600	43	91	8	0	16	Cubic	227 ± 64
600	54	115	6	0	22	Needles and cubic	
						1	

Too short residence times





Secoya Crystallization Technology

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Too short residence times

Polymorph selection and smallest size

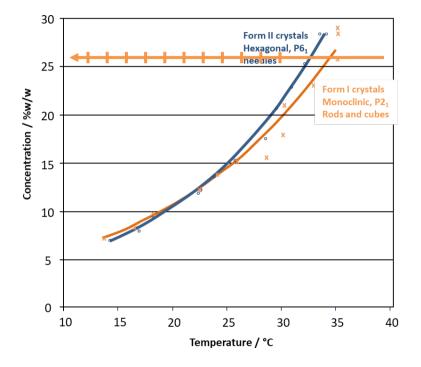


NH2



Secoya Crystallization Technology

And what about about shear rate



Thermodynamics:

- Start with heated solution
- Cool down to desired temperature where nucleation may take inside reactor with fixed dimension
- Let slurry obtained after passage at different tested nucleation temperatures grow to equilibrium
- Analyze crystal appearance and size as a function of tested condition

Hydrodynamics:

- Each solute in solution has an optimum average shear rate
- Other types of behavior?

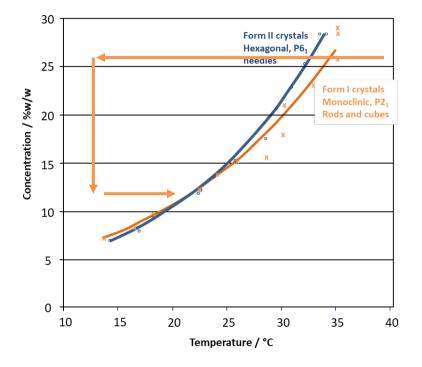


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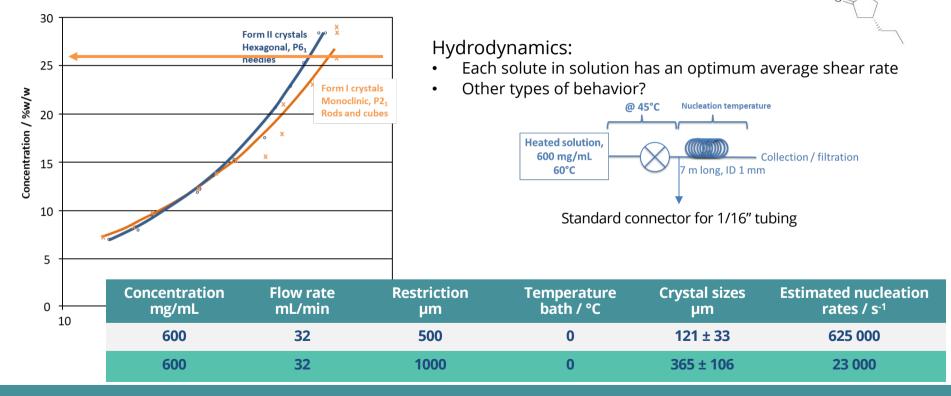
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C

NH2





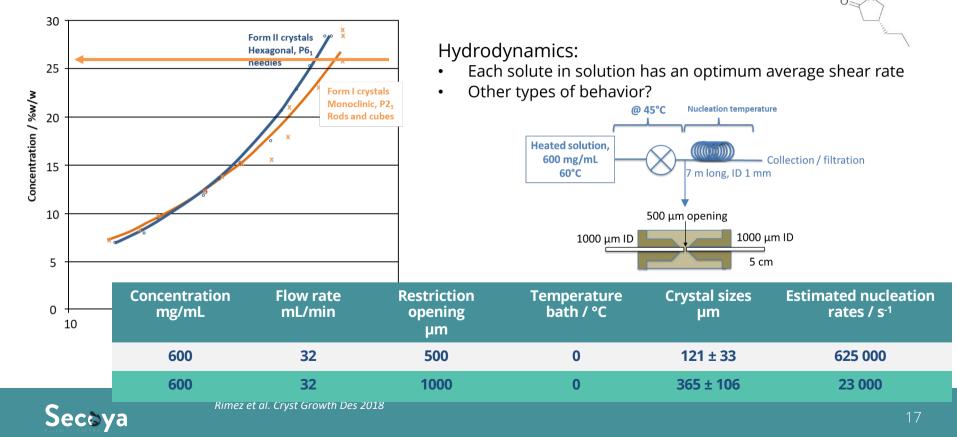


Rimez et al. Cryst Growth Des 2018

C

NH2



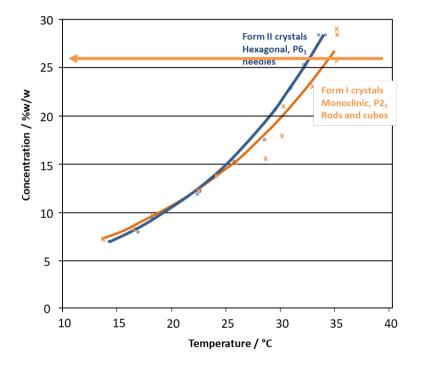


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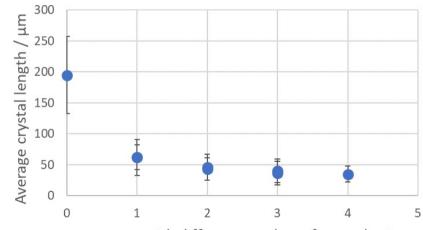
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And what about about shear rate



Hydrodynamics:

- Each solute in solution has an optimum average shear rate
- Repetitive placing of restrictions/openings/perturbations



test setup with different number of perturbations

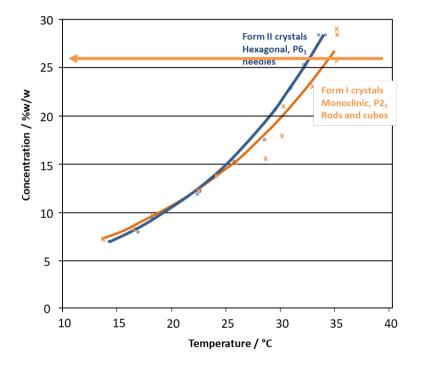


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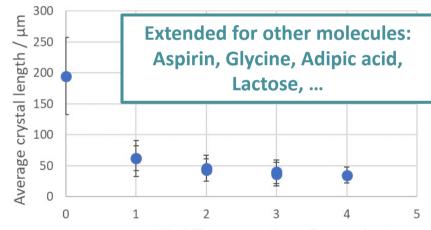
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test setup with different number of perturbations





- Finalizing acetylation step in tubular reactor
- Quench with water
- Crystallization in neat conditions
 acetic anhydride as solvent

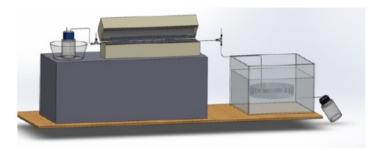
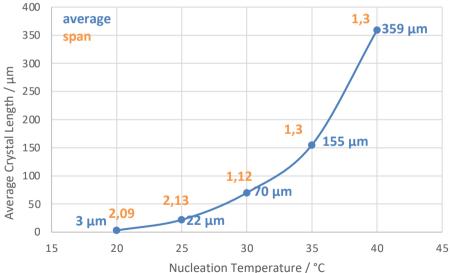


Fig. 7 Combined setup for the flow assisted synthesis and crystallisation of salicylic acid into aspirin, using identical connectors for sulphuric acid and water entry as shown in figure 1.



Aspirin

OF







- Finalizing acetylation step in tubular reactor .
- Quench with water
- Crystallization in neat conditions ٠ acetic anhydride as solvent

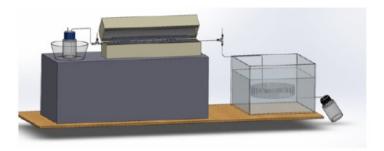
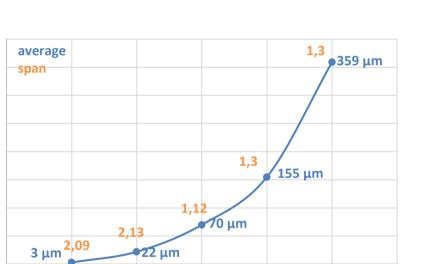


Fig. 7 Combined setup for the flow assisted synthesis and crystallisation of salicylic acid into aspirin, using identical connectors for sulphuric acid and water entry as shown in figure 1.



30

Nucleation Temperature / °C

35

40

22 um

25

400

350

300

250

200

150

100

50

0

15

20

Average Crystal Length / µm





Secoya

Rimez et al. React. Chem. Eng. 2019

45

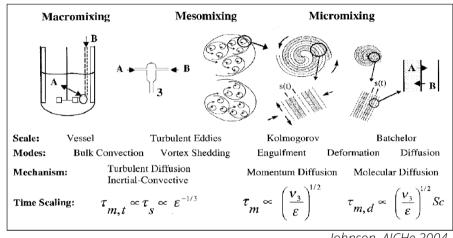


Secoya Crystallization Technology

Then we started to mix with antisolvents - now on purpose

Additional features with antisolvent

- Crystallization of organics at temperature between 0 and 70°C
- Precipitation of inorganic chemistries and salts: nanometric sizes
- Precipitation of APIs down to (sub-) micrometric sizes
- Co-crystallisation for enantiomeric purification project foreseen with partner



Johnson, AICHe 2004

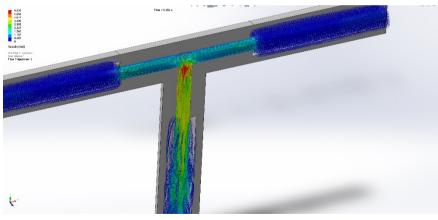
- mixing conditions without voids and dead volumes
- ✓ Mixing intensity determines nucleation rate
- ✓ Inherent small particle size



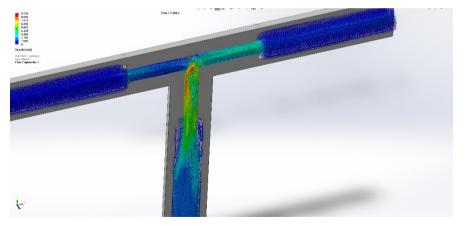


Frontal Collision mode: high energy mixing for equal quantities solvent/antsiolvent

20 and 20 mL/min



20 and 40 mL/min





Rimez et al. Cryst Growth Des. 2018



Secoya

Rimez et al. Cryst Growth Des. 2018

hexane

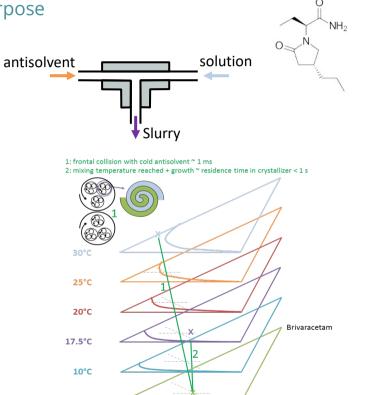
Secoya Crystallization Technology

Then we started to mix with antisolvents - now on purpose

Brivaracetam

- ✓ Difficult accessible chemistry and polymorphisms
- ✓ Excellent control of conditions

Crystallization	Initial concentration	Tubular	Mean	
method	API/IPAc	length		
	mg.mL ⁻¹	mm	μm	
Casted	150	500	10 ± 5	
	200	500	9 ± 5	
Filter paper	150	500	9 ± 4	
	200	500	7 ± 3	
	300	500	8 ± 3	



5°C IPAc

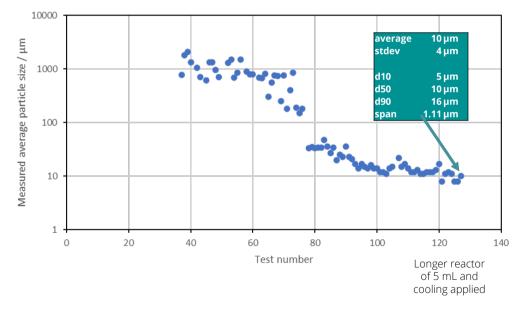
Brivaracetam



Paracetamol was dissolved in isopropanol (100 – 75 mg/mL) Water (unsuccesfull) and heptane were tested as antisolvent



- For paracetamol, 127 tests and 130 g solids on three solvent/antisolvent systems were necessary to arrive to particle sizes down to 10 μm
- SCT-LAB allows fast screening with low material consumption

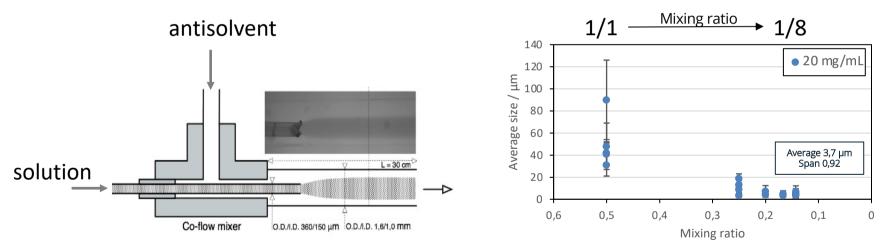






Actual customer example Steroid compound

Co-flow mixing mode: lowered energy of mixing, helps with increasing amount of antisolvent added



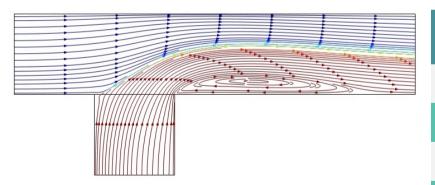
Down to 3 μ m, without the use of surfactant





side mixing mode: go in-between

Naproxen was dissolved in isopropanol (12.5 mg/mL) Water was used as antisolvent Sample collected in known quantity with surfactant (HPMC)





Product number	insert	Mixing ratio	Mn nm	MI nm	PDI
S27	Tside 500µm	1/3	910	2199	0.17
S28	Tside 500µm	1/4	30	581	0.42
S37	Tside 250 μm	1/3	183	658	0.09
S38	Tside 250µm	1/4	182	658	0.13





One Equipment that fits all needs

Obtain the crystal size of choice without hustle

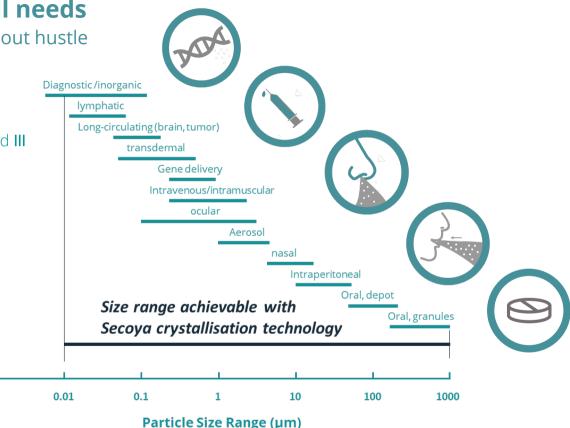
Cooling crystallization

Ideal for oral dosage forms Good soluble molecules – BCS Class I and III Optimised recycling and yield

- Antisolvent crystallization
 - Frontal co-flow side mix

Ideal solution for complex systems Reducing particle size under µm Adjustable particle size upon mixing conditions BCS Class II and IV

0.001

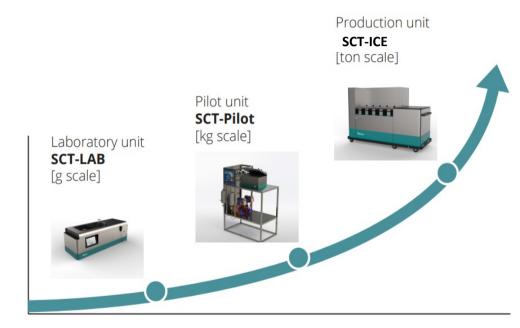






Technology specifications:

- 3 temperature zones:
 - Solution: RT to 85°C
 - Antisolvent: 5 to 85°C
 - Reactor: 0 to 70°C
- Delivered with dedicated cooling/heating thermostat
- Single use inserts and reactors
 - 6 different inserts for cooling and antisolvent crystallization
 - 6 different integrated reactors with different volumes: 1 to 7mL
 - $\circ~$ 1 specific reactor execution for highly viscous solutions
- Pump flow rates 1 to 60 mL/min
- Simplified collection of slurries
- Stand-alone 21 CFR part 11 software

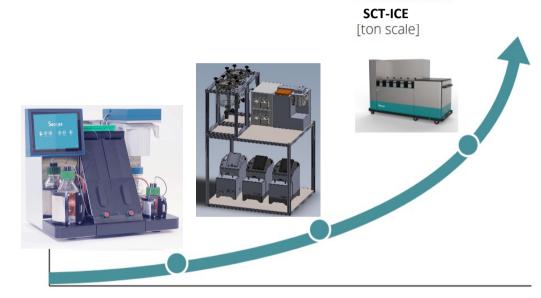






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Production unit

